

A NATURAL SECOBEYERENE

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The natural occurrence of the 3-carboxy-3,4-seco system has been reported for some of the common triterpene skeletons (1-4), however no examples of this structural feature appear to have been recorded in the diterpene series. We now describe the isolation and structural proof of 17-hydroxy-3,4-secobeyer-4(18),15-diene-3-oic acid (I).

Silicic acid chromatography of the acidic fraction of the ether extract of a Beyeria sp. nov. gave (I) as a resin, which was characterised as the methyl ester *p*-nitrobenzoate (II),  $C_{28}H_{35}O_6N$ , m.p.  $107-8^\circ$ ,  $[\alpha]_D + 45^\circ$ \*. Oxidation of (I) with the Jones reagent gave the diacid (IV),  $C_{20}H_{28}O_4$ , m.p.  $188-9^\circ$ ,  $[\alpha]_D + 46^\circ$ . Reduction of (I) or its methyl ester (III) with lithium aluminium hydride gave the diol (V),  $C_{20}H_{32}O_2$ , m.p.  $108-9^\circ$ ,  $[\alpha]_D + 9^\circ$ .

The methyl ester (III) showed I.R. absorption due to hydroxyl ( $3660\text{ cm}^{-1}$ ), ester carbonyl ( $1745\text{ cm}^{-1}$ ), terminal methylene ( $3070, 895\text{ cm}^{-1}$ ) and a *cis*-disubstituted olefinic group ( $3050, 750\text{ cm}^{-1}$ ). The 60 M/c n.m.r. spectrum ( $CDCl_3$ ) of (III) showed signals as singlets at  $0.80\ \delta$  (3H, tertiary methyl),  $1.77\ \delta$  (3H, vinylic methyl) and  $3.69\ \delta$  (3H, methoxyl). A singlet at  $3.52\ \delta$  (2H) which appeared at  $4.33\ \delta$  in the spectrum of the *p*-nitrobenzoate

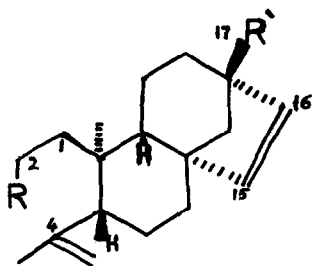
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\* Satisfactory microanalyses were obtained for all new compounds.  
Rotations were taken in  $CHCl_3$  except where otherwise stated.

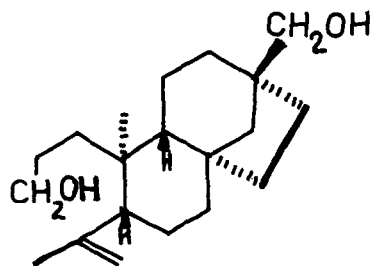
(II) was assigned to the resonance of the gem-protons of a primary carbinol attached to quaternary carbon. Resonances at 4.73  $\delta$  (1H) and 4.91  $\delta$  (1H) are consistent with a vinylidene group. An AB pattern ( $\delta_A$  5.82;  $\delta_B$  5.68;  $J_{AB}$  6 c.p.s.) corresponding to a cyclopentene in which the vinyl protons are coupled only to each other is expected for a beyer-15-ene system (5). Hydrogenation of the diol (V) using palladium-charcoal catalyst gave the dihydro derivative (VI),  $C_{20}H_{34}O_2$ , m.p. 123-4 $^\circ$ ,  $[\alpha]_D - 33^\circ$ , which lacked resonances attributed to the 15 and 16 vinylic protons. Further hydrogenation gave the tetrahydro derivative (VII),  $C_{20}H_{36}O_2$ , m.p. 137-8 $^\circ$ ,  $[\alpha]_D - 13^\circ$ , whose n.m.r. spectrum showed a triplet centred at 0.86  $\delta$  (6H) consistent with an isopropyl group. The diols (V), (VI) and (VII) all showed resonances of the  $C_3$  protons as triplets near 3.60  $\delta$  (2H;  $J = 6.5$  c.p.s.) requiring that the acid (I) be a monosubstituted acetic acid.

Confirmation of the structure (I) was obtained by cyclisation of the A ring and identification with the known methyl ester (IX) (6). For this purpose the diol (VI) was hydroborated and the product, m.p. 148-9 $^\circ$ , oxidized with excess Jones reagent. The acidic fraction thus obtained was heated with acetic anhydride and sublimed at 280 $^\circ$  to give the nor-keto acid (VIII),  $C_{19}H_{28}O_3$ , m.p. 184-6 $^\circ$ ,  $[\alpha]_D - 16^\circ$ , whose methyl ester was identical with an authentic sample. Alternatively, confirmation of the structure (I) could be achieved by A-ring fission following methods detailed for the triterpenes (1-4). Thus photolysis of beyeran-3-one (X) (7) in aqueous acetic acid gave (XI), characterised as the cyclohexylamine salt,  $C_{26}H_{47}O_2N$ , m.p. 145-6 $^\circ$ ,  $[\alpha]_D - 12^\circ$  (ethanol). Reduction of (XI) with LAH gave (XII) which formed a *p*-nitrobenzoate,  $C_{27}H_{39}O_4N$ , m.p. and mixed m.p. 110-111 $^\circ$ ,  $[\alpha]_D 0^\circ$ . The comparison sample of (XII) was obtained from the tosylate (XIII),  $C_{28}H_{38}O_5S$ , m.p. 87-8 $^\circ$ ,  $[\alpha]_D + 35^\circ$ , by reduction with LAH and hydrogenation.

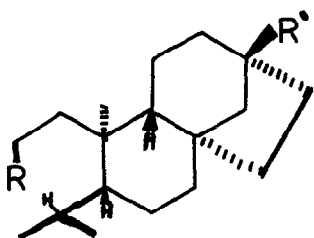
Full details of this and related work will be published in  
Aust. J. Chem.



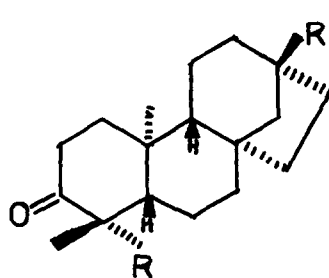
	R	R'
(I)	CO <sub>2</sub> H	CH <sub>2</sub> OH
(II)	CO <sub>2</sub> Me	CH <sub>2</sub> OCOC <sub>6</sub> H <sub>4</sub> NO <sub>2</sub>
(III)	CO <sub>2</sub> Me	CH <sub>2</sub> OH
(IV)	CO <sub>2</sub> H	CO <sub>2</sub> H
(V)	CH <sub>2</sub> OH	CH <sub>2</sub> OH
(XIII)	COOMe	CH <sub>2</sub> OTs



(VI)



	R	R'
(VII)	CH <sub>2</sub> OH	CH <sub>2</sub> OH
(XI)	COOH	CH <sub>3</sub>
(XII)	CH <sub>2</sub> OH	CH <sub>3</sub>



	R	R'
(VIII)	H	COOH
(IX)	H	CO <sub>2</sub> Me
(X)	CH <sub>3</sub>	CH <sub>3</sub>

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